Conjugate Addition to Thiin-4-ones; A Novel Route to Thiathromboxane Analogues

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<u>Summary</u>: In contrast to thiin-4-one, its 3-carbomethoxy-derivative undergoes efficient conjugate addition with organocuprate reagents; this procedure has been utilised as the key step in the synthesis of novel thiathromboxane analogues and in an extremely short formal synthesis of dithiathromboxane A_2 .

A wide range of α,β -unsaturated substrates undergo efficient organocuprate conjugate addition reactions¹ but, to our knowledge, such reactions have not been reported with thiin-4-one (4<u>H</u>-thiopyran-4-one) <u>1a</u> or its derivatives.^{2,3} We decided to investigate these reactions because, if successful, they could be employed to prepare a range of syntheticallyvaluable substituted dihydrothiin-4-ones⁴ and, by subsequent dehydrogenation, the corresponding substituted thiin-4-ones.⁵ In addition, this procedure also seemed well-suited⁶ to the synthesis of sulphur-containing analogues of Thromboxane A₂ (<u>2a</u>), e.g. dithiathromboxane A₂ (<u>2b</u>).⁷ Such compounds are of current interest in biological and pharmaceutical studies.⁸



Unfortunately, thiin-4-one $(\underline{1a})^3$ failed to react with a variety of organocopper and organocuprate reagents. However, we found that conjugate

addition reactions could be carried out if the corresponding 3-carbomethoxyderivative $(\underline{1b})^9$ was treated with lithium organocuprates in ether. Details of the scope and limitations of this reaction will be presented in a full paper; this letter illustrates the application of the methodology (i) to the synthesis of the novel thiathromboxane analogues $\underline{7}$, $\underline{8}$ and $\underline{9}$ and (ii) to an extremely short, formal synthesis of dithiathromboxane A_2 (<u>2b</u>), a compound with potent biological properties.⁷

The studies leading to the novel thiathromboxane analogues 7-9 are shown in Scheme 1. Treatment of thiopyran-4-one <u>1b</u> with cuprate <u>3</u> gave β keto ester <u>4</u>¹⁰ (82%) which was alkylated using NaH/THF followed by methyl 7-bromohept-5<u>Z</u>-enoate to give compound <u>5</u> in 62% yield. Demethoxycarbonylation of ester <u>5</u> proved somewhat troublesome but it was found that the transformation could be effected efficiently using magnesium chloride in aqueous dimethylformamide.¹¹ The required ketone <u>6</u> was obtained as a <u>cis-trans</u> mixture (<u>ca</u>. 40:60) in 69% yield along with a mixture of the corresponding alcohols <u>7</u> and <u>8</u> (16%). Desilylation (HF/CH₃CN) of <u>6</u> produced alcohols <u>7</u> and <u>8</u> in 92% yield and the two isomeric thiathromboxane analogues were separated using preparative centrifugal chromatography. The 15-diastereomers of both <u>7</u> and <u>8</u> were chromatographically identical in a number of solvent systems.

The thiopyranone thromboxane analogue <u>9</u> was also obtained from ketone <u>6</u> in 74% overall yield by dehydrogenation using <u>o</u>-chloranil followed by hydrofluoric acid desilylation.

The unsaturated thianone 6 was also employed to prepare a key precursor of dithiathromboxane A₂ (Scheme 2). Michael addition of methyl 3mercaptopropanoate gave a chromatographically-separable mixture of sulphide 11 and its 8,12-cis-epimer 10. Compound 10 was converted back to the mixture of enones 6 using sodium methoxide and the whole cycle repeated twice more to give a combined yield of 78% of the required epimer 11, the stereochemical assignments being supported by 400 MHz ¹H-n.m.r. We did not observe any of the corresponding 11β -adduct although this may have been present as a minor, co-chromatographing component. The stereoselective formation of the 11α -adduct <u>11</u> during thiol addition is to be expected from a consideration of the anomeric effect.^{7,12} Reduction of ketone 11 with sodium borohydride gave the required 9 β -alcohol 13 (31%) along with the 9 α epimer 12 (63%) which could be recycled by oxidation back to ketone 11 using pyridinium dichromate. Treatment of alcohol 13 with methanesulphonyl chloride and triethylamine produced mesylate 14 in 76% yield. It should be noted that desilylation took place during the work-up of the mesylation The ¹H-n.m.r. and i.r. spectra of mesylate <u>14</u> were essentially reaction. identical to those of an authentic sample.⁷ Mesylate 14 can be converted



 $\mathbb{R}^{1} = \underline{\mathbb{E}} - \mathbb{C} \mathbb{H} = \mathbb{C} \mathbb{H} : \mathbb{C} \mathbb{H} (\mathbb{O} \mathbb{S} \mathbb{I} \mathbb{M}_{2} \mathbb{B} \mathbb{u}^{t}) \mathbb{C}_{5} \mathbb{H}_{11} = \mathbb{R}^{2} = \underline{\mathbb{Z}} - \mathbb{C} \mathbb{H}_{2} \mathbb{C} \mathbb{H} = \mathbb{C} \mathbb{H} (\mathbb{C} \mathbb{H}_{2})_{3} \mathbb{C} \mathbb{O}_{2} \mathbb{M} \mathbb{E}$

ij.

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4

SCHEME 1

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5

into the sodium salt of dithiathromboxane A_2 (<u>2b</u>) in 4 steps using literature procedures.⁷ Overall, this methodology enables dithiathromboxane A_2 to be prepared in 10 steps from readily-available starting materials, under half the number used by the Ono group in their pioneering first synthesis.⁷

The biological properties of thiathromboxane analogues $\underline{7}$, $\underline{8}$ and $\underline{9}$ are currently being evaluated.

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